SCIENCE

PROGRAMME OF STUDY – Next Steps Medical Pathway – ST CHRISTOPHER'S ACADEMY

Prior learning: KS3

The principal focus of science teaching in key stage 3 is to develop a deeper understanding of a range of scientific ideas in the subject disciplines of biology, chemistry and physics. Pupils should begin to see the connections between these subject areas and become aware of some of the big ideas underpinning scientific knowledge and understanding. Examples of these big ideas are the links between structure and function in living organisms, the particulate model as the key to understanding the properties and interactions of matter in all its forms, and the resources and means of transfer of energy as key determinants of all these interactions. They should be encouraged to relate scientific explanations to phenomena in the world around them and start to use modelling and abstract ideas to develop and evaluate explanations.

Working Scientifically:	Biology:	Chemistry:	Physics:
investigations - Analysis and evaluation living organisms - Cells and organ - The skeletal and	living organisms - Cells and organisation - The skeletal and muscular systems - Nutrition and digestion	The particular nature of matter Atoms, elements and compounds Pure and impure substances	- Calculation of fuel uses and costs in the domestic context - Energy changes and transfers
	Gas exchange systemReproductionHealth	Chemical reactions Energetics	Motion and forces - Describing motion
	Material cycles and energy	The period table	ForcesPressure in fluidsBalanced forces
	PhotosynthesisCellular respiration	Materials Earth and Atmosphere	- Forces and motion Waves
			- Observed waves

	Interactions and interdependencies - Relationships in an ecosystem Genetics and evolution - Inheritance, chromosomes, DNA and genes	- Sound waves - Energy and waves - Light waves Electricity and electromagnetism - Current electricity - Static electricity - Magnetism Matter - Physical changes - Particle model - Energy in matter	
KS4	Science is changing our lives and is vital to the world's fut	Space Physics ture prosperity, and all students should be taught	
Taught Content: Knowledge/Skills	Science is changing our lives and is vital to the world's future prosperity, and all students should be taught essential aspects of the knowledge, methods, processes and uses of science. They should be helped to appreciate the achievements of science in showing how the complex and diverse phenomena of the natural world can be described in terms of several key ideas relating to the sciences which are inter-linked, and which are of universal application.		

These key ideas include:

- the use of conceptual models and theories to make sense of the observed diversity of natural phenomena
- the assumption that every effect has one or more cause
- that change is driven by interactions between different objects and systems
- that many such interactions occur over a distance and over time
- that science progresses through a cycle of hypothesis, practical experimentation, observation, theory development and review

• that quantitative analysis is a central element both of many theories and of scientific methods of inquiry.

This Programme of Study should comprise approximately equal proportions of biology, chemistry and physics. The relevant mathematical skills required are covered in the programme of study for mathematics and should be embedded in the science context.

Working Scientifically

Through the content across all three disciplines, students should be taught so that they develop understanding and first-hand experience of:

1. The development of scientific thinking

- the ways in which scientific methods and theories develop over time
- using a variety of concepts and models to develop scientific explanations and understanding
- appreciating the power and limitations of science and considering ethical issues which may arise
- explaining everyday and technological applications of science; evaluating associated personal, social, economic and environmental implications; and making decisions based on the evaluation of evidence and arguments
- evaluating risks both in practical science and the wider societal context, including perception of risk
- recognising the importance of peer review of results and of communication of results to a range of audiences.

2. Experimental skills and strategies

- using scientific theories and explanations to develop hypotheses
- planning experiments to make observations, test hypotheses or explore phenomena
- applying a knowledge of a range of techniques, apparatus, and materials to select those appropriate both for fieldwork and for experiments
- carrying out experiments appropriately, having due regard to the correct manipulation of apparatus, the accuracy of measurements and health and safety considerations
- recognising when to apply a knowledge of sampling techniques to ensure any samples collected are representative
- making and recording observations and measurements using a range of apparatus and methods
- evaluating methods and suggesting possible improvements and further investigations.

3. Analysis and evaluation

- applying the cycle of collecting, presenting and analysing data, including:
- presenting observations and other data using appropriate methods
- translating data from one form to another
- carrying out and representing mathematical and statistical analysis
- representing distributions of results and making estimations of uncertainty
- interpreting observations and other data, including identifying patterns and trends, making inferences and drawing conclusions
- presenting reasoned explanations, including relating data to hypotheses
- being objective, evaluating data in terms of accuracy, precision, repeatability and reproducibility and identifying potential sources of random and systematic error
- communicating the scientific rationale for investigations, including the methods used, the findings and reasoned conclusions, using paper-based and electronic reports and presentations.

4. Vocabulary, units, symbols and nomenclature

- developing their use of scientific vocabulary and nomenclature
- recognising the importance of scientific quantities and understanding how they are determined
- using SI units and IUPAC chemical nomenclature unless inappropriate
- using prefixes and powers of ten for orders of magnitude (e.g. tera, giga, mega, kilo, centi, milli, micro and nano)
- interconverting units
- using an appropriate number of significant figures in calculations.

Biology

Year 1

Year 2

Cell biology

- cells as the basic structural unit of all organisms; adaptations of cells related to their functions; the main sub-cellular structures of eukaryotic and prokaryotic cell
- Transport in cells; Diffusion; Osmosis and Active Transport
- enzymes
- factors affecting the rate of enzymatic reactions
- the importance of cellular respiration; the processes of aerobic and anaerobic respiration
- carbohydrates, proteins, nucleic acids and lipids as key biological molecules.

Transport systems

- the need for transport systems in multicellular organisms, including plants
- the relationship between the structure and functions of the human circulatory system.

Health, disease and the development of medicines

- the relationship between health and disease
- communicable diseases including sexually transmitted infections in humans (including HIV/AIDs) (taught through PSHE)

Coordination and control

- principles of nervous coordination and control in humans
- the relationship between the structure and function of the human nervous system
- the relationship between structure and function in a reflex arc
- principles of hormonal coordination and control in humans
- hormones in human reproduction, hormonal and non-hormonal methods of contraception
- homeostasis (blood glucose level only)

Photosynthesis

- photosynthesis as the key process for food production and therefore biomass for life
- the process of photosynthesis
- · factors affecting the rate of photosynthesis.

Chemistry

Atomic structure and the Periodic Table

- a simple model of the atom consisting of the nucleus and electrons, relative atomic mass, electronic charge and isotopes
- the number of particles in a given mass of a substance
- the modern Periodic Table, showing elements arranged in order of atomic number
- position of elements in the Periodic Table in relation to their atomic structure and arrangement of outer electrons
- properties and trends in properties of elements in the same group

- characteristic properties of metals and non-metals
- chemical reactivity of elements in relation to their position in the Periodic Table.

Structure, bonding and the properties of matter

- changes of state of matter in terms of particle kinetics, energy transfers and the relative strength of chemical bonds and intermolecular forces
- types of chemical bonding: ionic, covalent, and metallic
- bulk properties of materials related to bonding and intermolecular forces

Chemical changes

- determination of empirical formulae from the ratio of atoms of different kinds
- balanced chemical equations, ionic equations and state symbols
- identification of common gases
- the chemistry of acids; reactions with some metals and carbonates
- pH as a measure of hydrogen ion concentration and its numerical scale

Chemical analysis

- distinguishing between pure and impure substances
- separation techniques for mixtures of substances: filtration, crystallisation, chromatography, simple and fractional distillation
- quantitative interpretation of balanced equations

Physics

Energy

- energy changes in a system involving heating, doing work using forces, or doing work using an electric current: calculating the stored energies and energy changes involved
- power as the rate of transfer of energy
- conservation of energy in a closed system, dissipation
- calculating energy efficiency for any energy transfers
- renewable and non-renewable energy sources used on Earth, changes in how these are used.

Forces

	forces and fields: electrostatic, magnetic, gravity			
	• forces as vectors			
	 calculating work done as force x distance; elastic and inelastic stretching 			
	Forces and motion			
	 estimating speed and accelerations in everyday contexts 			
	 interpreting quantitatively graphs of distance, time, and speed 			
	• acceleration caused by forces; Newton's First Law			
	 weight and gravitational field strength 			
	 decelerations and braking distances involved on roads, safety 			
	Wave motion			
	• amplitude, wavelength, frequency, relating velocity to frequency and wavelength			
	• transverse and longitudinal waves			
	• electromagnetic waves, velocity in vacuum; waves transferring energy; wavelengths and frequencies from			
	radio to gamma-rays			
	• velocities differing between media: absorption, reflection, refraction effects			
	• production and detection, by electrical circuits, or by changes in atoms and nuclei			
	• uses in the radio, microwave, infra-red, visible, ultra-violet, X-ray and gamma ray regions, hazardous effects			
	on bodily tissues.			
Subsequent	Post 16+			
learning	This programme of study is designed to allow progression within other GCSE courses and to support			
	continuation and potential entry for GCSE or Level 1 or 2 qualifications at Dual Placement schools. This			
	programme of study could lead to supporting with skills required for a range of Level 2 and Level 3			
	qualifications or other vocational qualifications. It should support students to transition to adult life.			